



41

SC549.26TR

August 22, 1977

Improving the Analytical Accuracy of Electron Spin Resonance Spectrometry

Technical Report No. 10

Covering the Period July 1, 1977 to Aug. 20, 1977

Contract No. N00014-73-C-0325

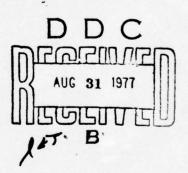
Task No. NR051-553

Prepared for

Office of Naval Research Arlington, VA. 22217

By

Ira B. Goldberg



Reproduction in whole or in part is permitted 'for any purpose of the United States government

Approved for public release; distribution unlimited





	READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER 2. GOVT ACCESSI	ION NO. 3. RECIPIENT'S CATALOG NUMBER
4. TITLE (end Subtitle) Improving the Analytical Accuracy of Electron Spin Resonance Spectroscopy.	5. TYPE OF REPORT & PERIOD COVE Technical Report 7/1/77 to 8/20/77
op in resonance spectroscopy?	14 SC549.26TR
Ira B./Goldberg	15 NØØØ14-73-C-Ø325
9. PERFORMING ORGANIZATION NAME AND ADDRESS Science Center	10. PROGRAM ELEMENT, PROJECT, TA
Rockwell International Thousand Oaks, CA 91360	NR 051-553
Materials Sciences Division Office of Naval Research	Aug 234 1977
Arlington, VA 22217 14. MONITORING AGENCY NAME & ADDRESS(It different from Controlling CO Office of Naval Research, Branch Office	Diffice) 15. SECURITY CLASS. (of this report)
1030 East Green St. Pasadena, CA 91106	Unclassified 15a. DECLASSIFICATION/DOWNGRADII SCHEDULE
Approved for Public Release, distribution unl Technical rept. no. 10, 1 Jul- 24 Aug 73,	ACCESSION for NTIS White Section
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if diffe	DDC Buff Section UNANNOUNCED JUSTIFICATION
	DISTRIBUTION/AVAILABILITY COOES
18. SUPPLEMENTARY NOTES	Dist. AVAIL. and/or SPEC

Sources of error in Analytical Electron Spin Resonance Spectrometry are discussed. Errors due to gain of the lock in amplifiers and modulation amplitude settings can be reduced if the settings are properly calibrated or if the attenuators are rebuilt. It is also necessary that the incident microwave power level is monitored and that the cavity is reproducibly tuned. A method of correction for incomplete integration of Lorentzian lines is discussed.

DD 1 JAN 73 1473

EDITION OF 1 NOV 65 IS OBSOLETE

389949

SECURITY CLASSIFICATION OF THIS PAGE (When Date Entered)

	GE(When Data Entered)
	mora see the common feets of the est subject
	ESSAGOA UKA BINAN POPTATINANTA ENGRAPAR
	fanotions at the state
The same of the sa	
	aseed spins of threezes (cost to soft)
RECEIVE OF THE STATE OF THE STA	
No see a see and	
to a see and	
No access along 2000	
NO SOURCE ALONG STORY OF THE ST	
NO SECURE SECURE DE SECURE DE LA CONTRACTOR DE LA CONTRAC	
NO SECULO	
A CONTRACTOR SECURITIES AND CONTRACTOR AND CONTRACT	
NO SECURE SECURE DE SECURE DE LA CONTRACTOR DE LA CONTRAC	
NA CONTRACTOR OF THE CONTRACTO	
ALL THE THE PROPERTY OF THE PR	
ALL THE THE PROPERTY OF THE PR	
A CONTRACTOR SECURITIES AND CONTRACTOR AND CONTRACT	The second of th
NA CONTRACTOR OF THE CONTRACTO	The second of th
A CONTRACTOR SECURITIES AND CONTRACTOR AND CONTRACT	The second of th
A CONTRACTOR SECURITIES AND CONTRACTOR AND CONTRACT	The second of th
STORY STORY STORY CONTROL OF THE STORY CONTROL OF T	A CONTRACT OF THE PROPERTY OF
N COM AND COM CANA COM AND COM AND COM AND COM	Total Terror Property of Color Property of Terror P
NO POST NEW CONTROL OF THE PARTY OF THE PART	paces while the street was the street with the street was a second or the s
N COME AND COME TO SERVICE OF THE PARTY OF T	A CONTROL OF THE PROPERTY OF T
N CONTROLL OF THE PROPERTY OF	ACTION OF THE PROPERTY OF THE
N CONTROLL OF THE PROPERTY OF	paces while the street was the street with the street was a street was



INTRODUCTION

Several reports (1-5) addressed the effects of instrumental parameters of electron spin resonance (ESR) spectrometers on the accuracy and precision of analytical determinations. Warren and Fitzgerald (2) carried out a systematic statistical evaluation of both instrumental and sampling parameters, and located many sources of random scattering of data. Nevertheless, the origin of some of these errors have not been elucidated with respect to improving the overall performance of analytical ESR spectroscopy.

Recently, we interfaced out ESR spectrometer to a laboratory computer (6) in order to study transient species as well as to carry out data reduction for analytical measurements. This has enabled us to examine the effect of instrumental parameters in a similar way to Warren and Fitzgerald (2), but with greater precision. From these experiments, we were able to determine several modifications to the instrument and to the operating procedure so that optimum performance of the instrument can be obtained. In this report, the causes of these instrumental errors are discussed, and steps are recommended to overcome these problems. These include the gain of the lock in amplifiers, modulation amplitude control, incident power level, and cavity tuning. Although our own instrument is a highly modified Varian V-4502 spectrometer, our results are generally applicable. Because many reported analytical measurements were carried out on the E-3 spectrometer (1-5), some of our results will be directed to that instrument.

EXPERIMENTAL SECTION

All measurements reported here were carried out on a modified Varian V-4502 ESR spectrometer with a dedicated PDP 8/m computer (6). The microwave power was monitored with a General Microwave, Model 476 power meter. Double



integrations were carried out subsequent to the experiments and corrected for finite integration length according to the method described in the appendix. Peak-to-peak amplitudes were measured as the difference between the maximum and minimum voltages of the digital spectrum and is accurate to ± 0.005 V relative to a maximum range of ± 0.5 V. Measurements of the modulation amplitude were also carried out as previously described (5) for our V-4502 spectrometer as well as two E-9 and two E-3 spectrometers.

RESULTS AND DISCUSSION

<u>Signal Level:</u> The output signal amplitude of the lock in amplifier is determined either by using a resistive divider to attenuate the signal, or to change the gain of one of the amplification stages prior to the output. For a signal divider, such as shown in Fig. 1, the attenuated voltage V_i divided by the signal input V_s into the attenuator, is given by equation 1.

$$f = \frac{V_{i}}{V_{S}} = \frac{\sum_{i=1}^{n} R_{i}}{\sum_{j=1}^{N} R_{j}}$$
 (1)

where the attenuated signal is measured across resistors i = 1 to n and the input signal is across resistors i = 1 to N. Given that resistors of a specified uncertainty, δR , are used, the error in the value of V_i/V_s is determined by the sum of the squares of the partial derivatives of V_i/V_s with respect to R_i 's and R_i 's (7). The result of this operation is

$$\frac{\delta(1/f)}{1/f} = \left\{ \frac{1}{\left(\sum_{i=1}^{n} R_{i}\right)^{2}} \cdot \left[\left(1 - \frac{1}{f}\right)^{2} \sum_{i=1}^{n} (\delta R_{i})^{2} + \sum_{j=1}^{N} (\delta R_{j})^{2} \right] \right\}^{\frac{1}{2}}$$
(2)



where $\delta(1/f)/(1/f)$ is equal to the relative error in the attenuation. For the Varian E-3, E-line series and V-4500 series spectrometers, this attenuator consists of two divider networks as shown in Fig. 1. Divider 1 provides steps of 25% in the attenuation while Divider 2 provides steps of a factor of 10. In the 100 KHz phase sensitive detectors of both the E-3 and V-4502 spectrometers, divider 1 is constructed of resistors of $\pm 5\%$ tolerance while divider 2 is constructed of resistors of $\pm 1\%$ tolerance. Both dividers on the E-line series are constructed of resistors of $\pm 1\%$ tolerance.

The effect of this design is shown in Table 1 for the E-3 spectrometer. Note that the gain indicated on the spectrometer and the gain calculated from the resistors deviate by as much as 5% (e.g., the indicated value of 1.25 is actually 1.31), and the uncertainty increases as the attenuation increases as predicted by Eq. 2. The situation is similar for the V-4502. We have investigated the gains on the V-4502 spectrometer utilizing actual ESR samples, a small 100 KHz voltage input to the phase sensitive detector, and a small DC voltage across the attenuator and find deviations of the magnitudes predicted by equations 1 and 2.

Divider 2 of the E-3 spectrometer has five gain settings. Those of 10^2 , 10^3 and 10^4 consist of three resistors shunted by the total resistance of divider 1. The gain settings of 10^5 and 10^6 alter the feedback loop of an amplifier. Thus, the three greatest amplifications contain a systematic error of $\pm 3.5\%$ determined principally by the total resistance of divider 1 and by the uncertainty of the feedback resistors.

The E-line series contains resistors of $\pm 1\%$ tolerance and is therefore more precise than either the V-4502 or E-3 spectrometers.

There are several solutions to correct for the uncertainty of the attenuation: the attentuator can either be calibrated or it can be rebuilt using



trimming resistors to make the attenuation values agree with the indicated values. We have chosen the latter approach since our instrument is used by a number of different people.

<u>Modulation Amplitude</u>: The modulation amplitude of the spectrometer is controlled through a resistive network similar to the signal level dividers. In addition the power of the output amplifier is directed into a low impedance 8Ω or 32Ω load. As a result there can be uncertainties in the modulation amplitude at the sample caused by the errors in the divider network as well as non-linearities at high modulation amplitude output powers.

The divider network for the divider 1 of the E-3, V-4502, and E-line spectrometers are identical to that of the signal level control shown in Fig. 1. Divider 2, which permits steps of 0.01, 0.1, 1.0, and 10 Gauss, is constructed of four resistors of ±1% tolerance. Errors for these are shown on Table 1 for the E-3 spectrometer. Similar results are obtained for the V-4502 spectrometer. The modulation amplitude attenuator of the newer E-line series is constructed from 1% resistors, and is thus subject to less uncertainty. As before, these errors can be rectified by proper calibration techniques or reconstruction. Since only relative values are important, calibration can be carried out on a sample with a linewidth much greater than the maximum modulation amplitude, such as MnSO4, such that the signal amplitude is linear with the modulation amplitude.

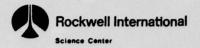
At modulation amplitudes greater than about 20 Oe, loading of the output amplifier can also contribute to deviations between the indicated and actual values. Randolph (8) has shown that the double integral of the ESR signal is proportional to the modulation amplitude. However, the amplitude of the derivative signal is only linear when the peak-to-peak modulation amplitude, $H_{\rm m}$,



is less than about 0.2 times the derivative peak-to-peak linewidth (9,10). At these values of $H_{\rm m}$ the sensitivity of the ESR spectrometer is much lower than optimum, such that higher modulation amplitudes are required for analysis of lower concentrations. Thus, in order to realize the full dynamic range of ESR, it may be necessary to calibrate at least the relative values of the higher modulation amplitudes. On the V-4502 spectrometer, we find that $H_{\rm m}$ is linear with the indicated value to 2500 "units" which corresponds to 12.5 Oe. Between 2500 "units" and 8000 "units", $H_{\rm m}$ is lower than expected, falling off to 29 Oe instead of the anticipated 40 Oe (11). At these high amplitudes, the waveform also becomes distorted.

<u>Power Level:</u> On the E-3, E-4 and V-4500 spectrometers, the power level is determined by a vane or flap attenuator in which a lossy material is inserted into the waveguide. The maximum accuracy of these attenuators is ± 0.5 db to about 40 db and ± 1 db to about 60 db. The reproducibility is only slightly better. Thus, depending on the range of attenuation this can amount to a $\pm 6\%$ to $\pm 12\%$ error in the power level. The rotary vane attenuators used in the E-line series offer slightly greater accuracy but much greater reproducibility. Nevertheless, it is advisable to utilize a power meter to monitor the incident power to the cavity. These are reproducible to $\pm 0.5\%$ and accurate to $\pm 1\%$. This permits materials which require different incident powers to be compared to any given standard. We make the assumption here that the cavity tuning remains fixed as does the bias level to the detector.

<u>Cavity Tuning:</u> One of the most critical aspects of reproducible ESR determinations is that the cavity is reproducibly tuned. In addition, the detector should be biassed at the same operating point. Often, once the microwave power is determined, the iris on the cavity is adjusted such that the



detector is biassed at the same current level for all measurements. This however precludes the cavity being tuned in a reproducible manner. Alternatively, microwave power can be supplied to the detector in which this biasing radiation is in phase with the radiation reflected from the microwave cavity. This permits the cavity to be maintained near critical coupling which is the most sensitive and most reproducibly set point. Results of these methods are illustrated in Fig. 2. At incident powers greater than about 10mW (c.a. 17 db), the point of critical coupling is very close to the point at which sufficient power is reflected to properly bias the detector, and the ESR performance is equal to that of the method of external biassing. Below that value, the cavity ris must be opened which decreases the power density at the sample. Thus, the value of $A/P^{\frac{1}{2}}$, where A is the peak to peak amplitude of the derivative of the absorption and P is the incident power, decreases as the attenuation is increased. If biassing is carried out using an external arrangement, such as provided on the E-line series spectrometers, the value of A/P½ remains constant as long as the resonance transition is not saturated. A plot of log A vs log P, for a sample of powdered $MnSO_4 \cdot H_2O$ gave a least-squares slope of 0.503 ± 0.006 over range of 60 db attenuation.

The results shown here agree with those presented by Warren and Fitzgerald (2), and explain some of the errors of ESR determinations. High precision can be obtained if the properties of the detection system and microwave cavity are taken into consideration. As others have concluded (2,4), changing the modulation amplitude introduces additional uncertainty into the measurements. However, this uncertainty is not so great if the modulation amplitude is calibrated and the non-linearity of the peak-to-peak amplitude with modulation amplitude is taken into account (1,5), or if the double integral is used as a



measure of concentration (1,5,8). We assume that the samples are not saturated by high incident powers (12), and that the cavity is not loaded by excessive material (13).

ACKNOWLEDGEMENTS

We thank Jean Kertesz for providing schematic drawings of the E-3 spectrometer, and N. Pchelkin, George Schneider and Allen Bard for the use of their Spectrometers.

This work was supported in part by the Office of Naval Research.



Table I

Factors for Signal Level and Modulation Amplitude^a and Uncertainties Associated with These Factors for an E-3 Spectrometer

indicated gain	theorecical ^C gain	uncertainty ^d (%)	attentuation factor	
Signal Level and Modulation				
0.5 0.63 0.8 1.0 1.25 1.6 2.0 2.5 3.2 4.0	0.509 0.608 0.808 1.007 1.307 1.606 1.995 2.504 3.182 4.000	4.7 4.0 3.2 2.7 2.3 2.1 1.8 1.5 1.1	0.13 0.15 0.20 0.25 0.33 0.40 0.50 0.63 0.80	
Signal Level				
1.0x10 ² 1.0x10 ³ 1.0x10 ⁴ 1.0x10 ⁵ 1.0x10 ⁶	1.003×10 ² 0.990×10 ³ 1.008×10 ⁴ 1.017×10 ⁵ 1.047×10 ⁶	1.3 1.1 3.5b 3.5+1.0b 3.5+0.9b	0.01 0.1 1.0 10.	
Modulation Amplitude				
0.01 0.1 1.0 10.0	0.01003 0.0990 1.008 10.00	1.3 1.3 3.6 0.0	1.0×10 ⁻³ 1.0×10 ⁻² 1.0×10 ⁻¹ 1.0	

^a determined from E-3 schematics

 $^{^{\}mathrm{b}}$ for these two settings, 3.5% represents a systematic error

c calculated from equation 1

d calculation from equation 2



Appendix 1. Double Integration of the Derivative of Lorentzian Absorption Lines

The derivative of Lorentzian absorption lines, normalized to unity, is described by equation A-1 (14)

$$Y' = \frac{1}{\pi \Gamma} \cdot \frac{2\Gamma^2 (H-H_0)}{[\Gamma^2 + (H-H_0)^2]^2}$$
 (A-1)

where H is the applied magnetic field, H_0 is the magnetic field at resonance and Γ is the half-width of the absorption at half-height. The double integral of this lineshape between the points H_0 - H_a and H_0 + H_a is then given by

$$I = \int_{H_0 - H_a}^{H_0 + H_a} \int_{H_0 - H_a}^{H} Y' dH' dH = \frac{2}{\pi} \left\{ tan^{-1} \left[\frac{(H_0 - H_a)}{\Gamma} \right] - \frac{\frac{(H_i - H_a)}{\Gamma}}{1 + \left(\frac{H_0 - H_a}{\Gamma} \right)^2} \right\}$$
(A-2)

Assuming that $\triangle H_{pp} = \frac{\sqrt{2}}{3}$, and dividing the experimentally determined double integral by I, gives the value of the double integral which would be obtained if an infinite scan could be used.

The computer program used for integration defines the field at maximum absorption and computes the peak to peak width. Magnetic fields equidistant from the absorption maximum are then defined, and the zero-derivative point is determined. Double integration is then carried out by Simpson's rule, and the correction factor described by equation A-2 is then applied to the results. A series of twenty experiments on a sample of $MnSO_4 \cdot H_2O$ yielded a standard deviation of 0.6% of the mean value.

This approach can also be used for multiple overlapping lines. Typically, only the outermost lines need to be treated. Because many lines are contained in the spectrum, small errors in the correction factor do not greatly influence the final result.



- M. L. Randolph "Quantitative Considerations in Electron Spin Resonance of Biological Materials", in H. M. Swartz, J. R. Bolton, and D. C. Borg, eds. <u>Biological Applications of Electron Spin Resonance</u>, J. Wiley, N. Y., 1972 Chapter 3.
- 2) D. C. Warren and J. M. Fitzgerald, Anal. Chem., 49, 250 (1977).
- W. G. Bryson, D. P. Hubbard, B. M. Peake, and J. Simpson, Anal. Chim Acta, 77, 107 (1975).
- 4) G. C. Guilbault and G. J. Lubrano, Anal. Lett., 1, 725 (1968).
- I. B. Goldberg and A. J. Bard, "Electron Spin Resonance", M. M. Bursey,
 I. M. Kolthoff and P. J. Elving, eds "Treatise on Analytical Chemistry",
 2nd ed., in press.
- I. B. Goldberg, H. R. Crowe, and R. S. Carpenter II, J. Magn. Reson. <u>18</u>, 84 (1975).
- 7) P. R. Bevington, <u>Data Reduction and Error Analysis for the Physical Sciences</u>, McGraw-Hill, New York, 1969, Chapter 4.
- 8) M. L. Randolph, Rev. Sci Instrum., 39, 149 (1960).
- 9) G. W. Smith, J. Appl. Phys., <u>35</u>, 1217 (1964).
- 10) H. Walhurst, J. Chem. Phys., <u>35</u>, 1708 (1961)
- I. B. Goldberg, A. J. Lewin, and J. R. Crandall, Rev. Sci. Instrum., <u>45</u>, 855 (1974).
- 12) C. Mailer, T. Sarna, H. M. Swartz, and J. S. Hyde, J. Magn. Reson., <u>25</u>, 205 (1977).
- 13) I. B. Goldberg and H. R. Crowe, Anal. Chem., <u>49</u>, 1353 (1977).
- 14) J. E. Wertz and J. R. Bolton, <u>Electron Spin Resonance</u>; <u>Elementary Theory</u> and <u>Practical Applications</u>; <u>McGraw-Hill</u>, N. Y. 1972, pp. 32-36.

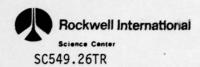


Figure 1. Schematic representation of the attenuator used in the V-4502 ESR spectrometer to control the output signal level and modulation amplitude of the 100 KHz phase sensitive detector. Resistance values are given below the resistor in ohms, and the attenuation normalized to a value of 8000 is given adjacent to the contact points.

Figure 2. The ESR signal for strong pitch divided by the square root of the power incident to the microwave power vs the attenuation of microwave power. O db corresponds to 580 mW. Power saturation of the sample begins at attenuations small than 20 db.

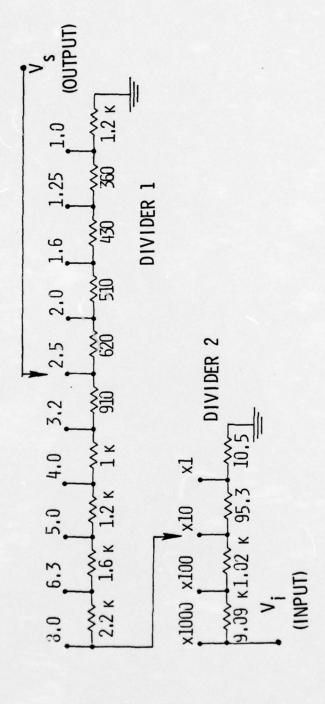
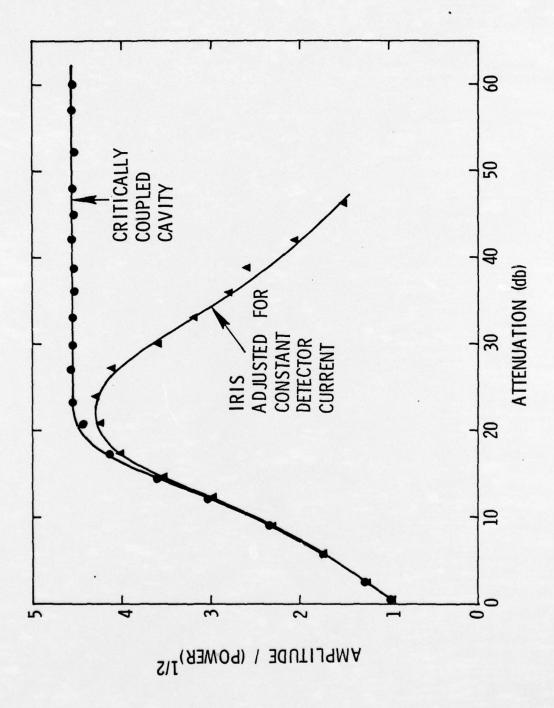


Figure 1

Figure 2



13

TECHNICAL REPORT DISTRIBUTION LIST

	TECHNICAL REPORT	DISTRIBUTION LIST	
	No. Copies		No. Copie
Dr. M. B. Denton University of Arizona Department of Chemistry Tucson, Arizona 85721	1	Dr. Fred Saalfeld Naval Research Laboratory Code 6110 Washington, D.C. 20375	1
Dr. G. S. Wilson University of Arizona Department of Chemistry Tucson, Arizona 85721	1	Dr. H. Chernoff Massachusetts Institute of Tec Department of Mathematics Cambridge, Massachusetts 02139	
Dr. R. A. Osteryoung Coloredo State University Department of Chemistry Fort Collins, Colorado 80521	1	Dr. K. Wilson University of California, San Department of Chemistry La Jolla, California 92037	Diego 1
Dr. B. R. Kowalski University of Washington Department of Chemistry Seattle, Washington 98105	1	Dr. A. Zirino Naval Undersea Center San Diego, California 92132	1
Dr. I. B. Goldberg North American Rockwell Scients. P.O. Box 1085	nee Center	Dr. John Duffin United States Naval Post Gradu Monterey, California 93940	ate School
1049 Camino Dos Rios Thousand Caks, California 91 Dr. S. P. Perone Purdue University	360 1	Dr. G. M. Hieftje Department of Chemistry Indiana University Bloomington, Indiana 47401	1
Department of Chemistry Lafayette, Indiana 47907 Dr. E. E. Wells	1	Dr. Victor L. Rehn Naval Weapons Center Code China Laka California 03555	1
Naval Research Laboratory Code 6160 Washington, D.C. 20375	1	China Lake, California 93555	1
Dr. D. L. Venezky Naval Research Laboratory Code 6130			,
Washington, D.C. 20375	1		

1

Dr. H. Freiser University of Arizona Department of Chemistry Tuscon, Arizona 85721

TECHNICAL REPORT DISTRIBUTION LIST

• • •	TECHNICAL REPOR	T DISTRIBUTION LIST	
	No. Copies	<u>N</u>	o. Copi
Office of Naval Research Arlington, Virginia 22217 Attn: Code 472	2	Defense Documentation Center Building 5, Cameron Station Alexandria, Virginia 22314	12
Office of Naval Research Arlington, Virginia 22217 Attn: Code 102IP	6	U.S. Army Research Office P.O. Box 12211 Research Triangle Park, North Carolin Attn: CRD-AA-IP	a 27709
ONR Branch Office 536 S. Clark Street Chicago, Illinois 60605 Attn: Dr. George Sandoz ONR Branch Office	. 1	Commander Naval Undersea Research & Development Center San Diego, California 92132 Attn: Technical Library, Code 133	1
715 Broadway New York, New York 10003 Attn: Scientific Dept. ONR Branch Office	1	Naval Weapons Center China Lake, California 93555 Attn: Head, Chemistry Division	1
1030 East Green Street Pasadena, California 91106 Attn: Dr. R. J. Marcus	1	Naval Civil Engineering Laboratory Port Hueneme, California 93041 Attn: Mr. W. S. Haynes	1
Office of Naval Research San Francisco Area Office One Hallidie Plaza - Suite 60 San Francisco, CA 94102 Attn: Dr. P. A. Miller	1	Professor O. Heinz Department of Physics & Chemistry Naval Postgraduate School Monterey, California 93940	1
ONR Branch Office 495 Summer Street Boston, Massachusetts 02210 Attn: Dr. L. H. Peebles	1	Dr. A. L. Slafkosky Scientific Advisor Commandant of the Marine Corps (Code : Washington, D.C. 20380	RD-1)
Director, Naval Research Labor Washington, D.C. 20390 Attn: Library, Code 2029 (ONE Technical Info. Div. Code 6100, 6170 The Asst. Secretary of the Nav	RL) 6 1 1		

Department of the Navy Room 4E736, Pentagon Washington, D.C. 20350

Commander, Naval Air Systems Command

Department of the Navy Washington, D.C. 20360 Attn: Code 310C (H. Rosenwasser) 1





